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Lab report on copper cycle - slideshare

Oct 28, 2013 References: Lab manual: Koczanski, Krystyna; Xidos, James D. CHEM 1300 Laboratory Manual; UMSU Copy Centre: Presentation On Copper, Cu cthame1. 7,605

Using a table of standard reduction - yahoo

Apr 09, 2008 Best Answer: $\text{Cu(s)} \rightarrow \text{Cu}^{2+}(\text{aq})$ $E_o = -0.337$ volts $\text{Cl}_2(\text{g}) + 2\text{Cl}^{-}(\text{aq})$ $E_o = 1.359$ volts _____ $\text{Cu(s)} + \text{Cl}_2(\text{g}) + \text{Cu}^{2+}(\text{aq}) + 2\text{Cl}^{-}(\text{aq})$ $E_o = 1.022$ volts

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Jul 28, 2006 and temperature upon systems at equilibrium. PRE-LAB EXERCISES Do not answer the questions in the laboratory manual at this time $[\text{Cu}(\text{NH}_3)_4]^{2+}$