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Most chemical elements are metals. They are located in the left side and in the middle of the table. Their properties include large atomic radius, low ionization  
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Combining metals with different properties . they are so reactive because they have only one valence electron in their Properties of Elements Study Guide  
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based on the chemical and physical properties of elements, into triads based on their chemical properties answer to this question is thought  
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### **(a) classify the following elements as metals or**

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The periodic table we use today Similar chemical properties to other elements in because he predicted their existence and their properties before their

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and each answer can be used more than once. CHAPTER 20 ELEMENTS AND THEIR PROPERTIES Author: Donor Last modified by: Donor Created Date: 3/22/2010 2:52:00 AM

### **What are the properties of group 2 elements? |**

Sep 10, 2008 Best Answer: Here goes: Group 2 The chemical properties of Group 2 elements are dominated by the strong In all their compounds these metals have

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students compare the physical and chemical properties of elements by their They complete a worksheet and compare answers with their A Reading Guide

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Mar 16, 2010 Best Answer: Their constituent atoms, What gives elements their properties? what gives elements their properties. Add your answer. Source.

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Nov 02, 2009 Why can elements be arranged in rows and columns by their properties? Best Answer: It's In other words what makes elements have properties in

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